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## Low-temperature redetermination of 3,4,5,6-tetrahydropyrimidin-2(1H)-one

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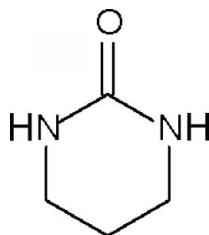
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Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.001$  Å;  $R$  factor = 0.037;  $wR$  factor = 0.111; data-to-parameter ratio = 13.6.

The low-temperature structure of the title compound,  $\text{C}_4\text{H}_8\text{N}_2\text{O}$ , is ordered, whereas the central methylene groups is disordered in the reported room-temperature structure. The molecule lies across a mirror plane; adjacent molecules are linked by an  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond into a chain.

### Related literature

For the room-temperature, disordered structure of tetrahydropyrimidin-2(1H)-one, see: Calogero *et al.* (1980).



### Experimental

#### Crystal data

$\text{C}_4\text{H}_8\text{N}_2\text{O}$   
 $M_r = 100.12$

Orthorhombic,  $Pnma$   
 $a = 9.9958$  (1) Å

$b = 7.1327$  (1) Å  
 $c = 6.7365$  (1) Å  
 $V = 480.29$  (1) Å<sup>3</sup>  
 $Z = 4$

Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 100$  (2) K  
 $0.35 \times 0.20 \times 0.15$  mm

#### Data collection

Bruker SMART APEX  
diffractometer  
Absorption correction: none  
6595 measured reflections

749 independent reflections  
719 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.110$   
 $S = 1.06$   
749 reflections

55 parameters  
All H-atom parameters refined  
 $\Delta\rho_{\text{max}} = 0.49$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.89 (1)	1.97 (1)	2.864 (1)	178 (1)

Symmetry code: (i)  $-x + 1, -y + 1, -z + 1$ .

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

We thank the University of Malaya for the purchase of the diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BV2094).

### References

- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.  
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Calogero, S., Russo, U. & Del Pra, A. (1980). *J. Chem. Soc. Dalton Trans.* pp. 646–653.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
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**supplementary materials**

*Acta Cryst.* (2008). E64, o914 [ doi:10.1107/S160053680801115X ]

## Low-temperature redetermination of 3,4,5,6-tetrahydropyrimidin-2(1*H*)-one

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### Comment

The crystal structure of tetrahydropyrimidin-2(1*H*)-one (Scheme I) was refined as a disorder model, the second of the three methylene carbon atoms being disordered about a mirror plane by a ratio of 0.7:0.3 (Calogero *et al.*, 1980). However, the structure is ordered at low temperature (Fig. 1); adjacent molecules are linked by a N–H···O hydrogen bond into a chain motif.

### Experimental

The commercially available compound was crystalline. A large block was cut into a smaller specimen.

### Refinement

All hydrogen atoms were located in a difference Fourier map, and were freely refined.

### Figures

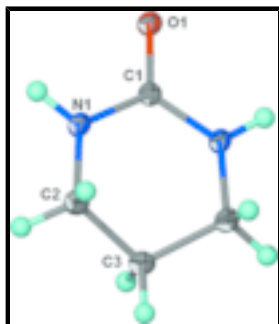


Fig. 1. Thermal ellipsoid plot of tetrahydropyrimidin-2(1*H*)-one at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

## 3,4,5,6-tetrahydropyrimidin-2(1*H*)-one

### Crystal data

C<sub>4</sub>H<sub>8</sub>N<sub>2</sub>O

*M<sub>r</sub>* = 100.12

Orthorhombic, *Pnma*

Hall symbol: -P 2ac 2n

*a* = 9.9958 (1) Å

*b* = 7.1327 (1) Å

*c* = 6.7365 (1) Å

*V* = 480.29 (1) Å<sup>3</sup>

*F*<sub>000</sub> = 216

*D<sub>x</sub>* = 1.385 Mg m<sup>-3</sup>

Mo *K*α radiation

λ = 0.71073 Å

Cell parameters from 4972 reflections

θ = 3.0–30.4°

μ = 0.10 mm<sup>-1</sup>

*T* = 100 (2) K

Block, colorless

# supplementary materials

Z = 4

0.35 × 0.20 × 0.15 mm

## Data collection

Bruker SMART APEXII diffractometer	719 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.020$
Monochromator: graphite	$\theta_{\text{max}} = 30.0^\circ$
$T = 100(2)$ K	$\theta_{\text{min}} = 3.7^\circ$
$\varphi$ and $\omega$ scans	$h = -12 \rightarrow 13$
Absorption correction: none	$k = -9 \rightarrow 10$
6595 measured reflections	$l = -9 \rightarrow 9$
749 independent reflections	

## Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.036$	All H-atom parameters refined
$wR(F^2) = 0.110$	$w = 1/[\sigma^2(F_o^2) + (0.0818P)^2 + 0.084P]$
$S = 1.06$	where $P = (F_o^2 + 2F_c^2)/3$
749 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
55 parameters	$\Delta\rho_{\text{max}} = 0.49 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$
	Extinction correction: none

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.48291 (8)	0.7500	0.51927 (11)	0.0143 (2)
N1	0.58480 (7)	0.58713 (8)	0.27287 (9)	0.0140 (2)
C1	0.54996 (10)	0.7500	0.36064 (14)	0.0113 (2)
C2	0.67121 (7)	0.57605 (10)	0.09858 (11)	0.0145 (2)
C3	0.65258 (11)	0.7500	-0.02874 (14)	0.0146 (3)
H1	0.5642 (13)	0.4836 (18)	0.3405 (19)	0.023 (3)*
H21	0.6483 (11)	0.4643 (17)	0.0232 (17)	0.017 (3)*
H22	0.7668 (11)	0.5621 (17)	0.1397 (17)	0.017 (2)*
H31	0.5609 (18)	0.7500	-0.089 (3)	0.023 (4)*
H32	0.7163 (18)	0.7500	-0.140 (3)	0.023 (4)*

## Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0176 (4)	0.0122 (4)	0.0132 (4)	0.000	0.0045 (3)	0.000
N1	0.0188 (4)	0.0099 (3)	0.0133 (3)	0.0003 (2)	0.0047 (2)	0.00012 (19)
C1	0.0109 (4)	0.0111 (5)	0.0118 (4)	0.000	-0.0008 (3)	0.000

C2	0.0173 (4)	0.0128 (4)	0.0133 (4)	0.0010 (2)	0.0039 (2)	-0.0008 (2)
C3	0.0177 (5)	0.0145 (5)	0.0117 (4)	0.000	0.0013 (3)	0.000

*Geometric parameters (Å, °)*

O1—C1	1.2614 (12)	C2—H21	0.972 (12)
N1—C1	1.3492 (8)	C2—H22	1.000 (11)
N1—C2	1.4597 (9)	C3—C2 <sup>i</sup>	1.5198 (9)
N1—H1	0.892 (13)	C3—H31	1.001 (18)
C1—N1 <sup>i</sup>	1.3492 (8)	C3—H32	0.982 (19)
C2—C3	1.5198 (9)		
C1—N1—C2	123.49 (6)	N1—C2—H22	110.4 (7)
C1—N1—H1	115.4 (8)	C3—C2—H22	110.8 (7)
C2—N1—H1	120.2 (8)	H21—C2—H22	106.8 (10)
O1—C1—N1	120.56 (4)	C2 <sup>i</sup> —C3—C2	109.45 (8)
O1—C1—N1 <sup>i</sup>	120.56 (4)	C2 <sup>i</sup> —C3—H31	109.9 (5)
N1—C1—N1 <sup>i</sup>	118.86 (9)	C2—C3—H31	109.9 (5)
N1—C2—C3	109.71 (6)	C2 <sup>i</sup> —C3—H32	110.5 (5)
N1—C2—H21	109.0 (7)	C2—C3—H32	110.5 (5)
C3—C2—H21	110.2 (7)	H31—C3—H32	106.7 (15)
C2—N1—C1—O1	-174.85 (8)	C1—N1—C2—C3	-30.95 (10)
C2—N1—C1—N1 <sup>i</sup>	7.00 (14)	N1—C2—C3—C2 <sup>i</sup>	52.71 (10)

Symmetry codes: (i)  $x, -y+3/2, z$ .

*Hydrogen-bond geometry (Å, °)*

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 $\cdots$ O1 <sup>ii</sup>	0.89 (1)	1.97 (1)	2.864 (1)	178 (1)

Symmetry codes: (ii)  $-x+1, -y+1, -z+1$ .

Fig. 1

